

Figure 1: Variation of Equilibrium Conditional Solubility versus pH for Struvite (from Ohlinger et al., 1998)

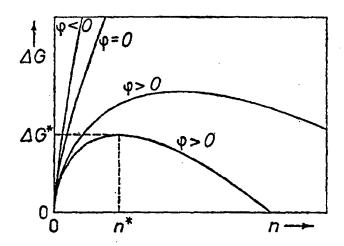


Figure 2: Free Energy versus Number of Particles in a Precipitating Crystal

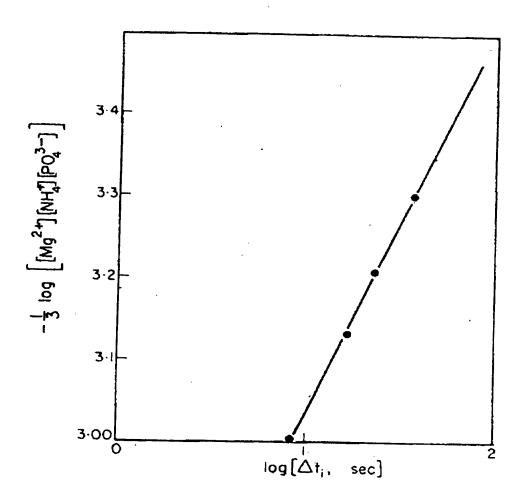


Figure 3: Concentration (-1/3 log of Ionic Product, Mol/L) versus Induction Time (Sec) for

Struvite Precipitation

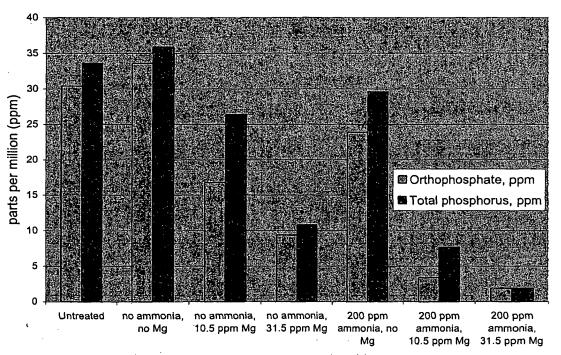


Figure 4: Dissolved OP and TP (ppm) in Untreated and Treated Effluent from Rocky Mount Lagoon

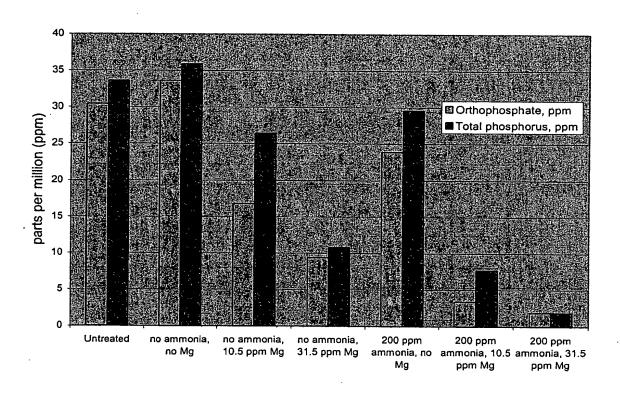


Figure 5: Dissolved OP and TP (ppm) in Untreated and Treated Effluent from Clayton Digester

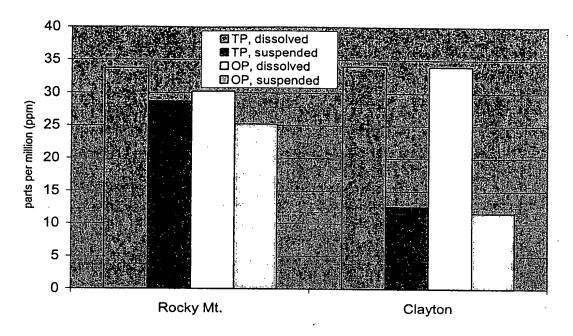


Figure 6: Breakdown of Phosphorus Content (ppm) by Form in Rocky Mount and Clayton Effluent

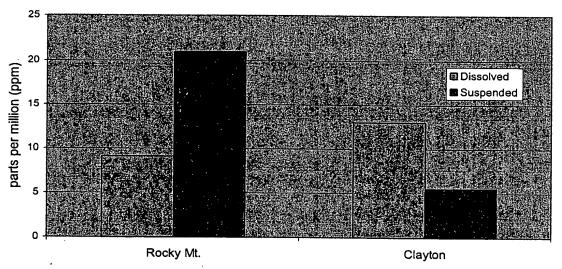


Figure 7: Breakdown of Mg Content (ppm) by Form in Rocky Mount and Clayton Effluents

Title: APPARATUS AND METHOD FOR REMOVING PHOSPHORUS FROM WASTE LAGOON EFFLUENT Applicant(s): Bowers et al. Atty. Docket No.: 297/181

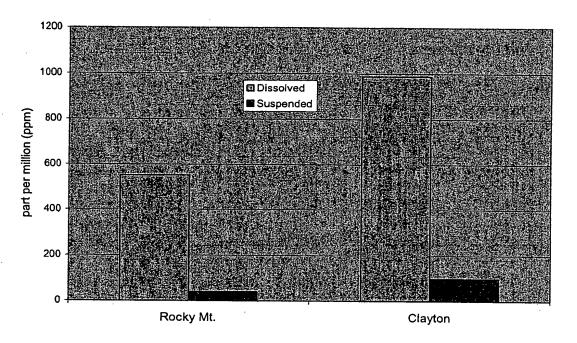


Figure 8: Breakdown of TAN (ppm) by Form in Rocky Mount and Clayton Effluent

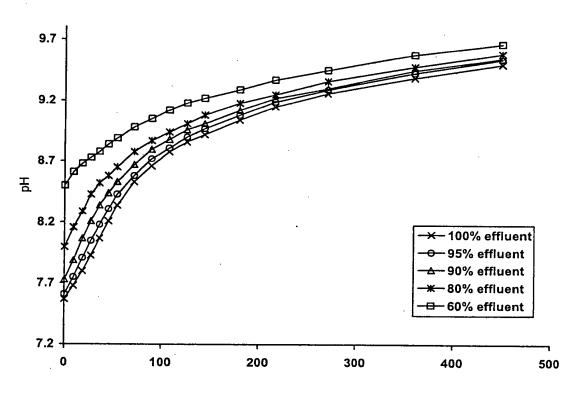


Figure 9: pH versus Amount of Ammonia Added (ppm) for Five Ratios of Effluent to Mg-Supplementing Solution

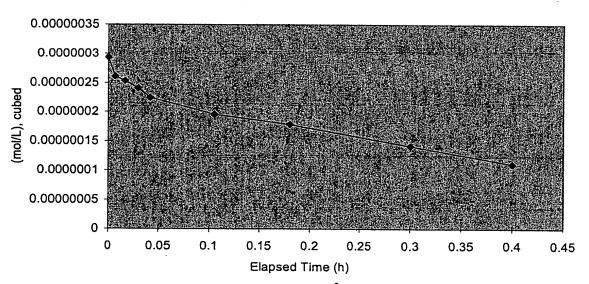


Figure 10: Excess Molar Product (mol/L)³ versus Time (h) Elapsed from pH, OP, and Mg Augmentation

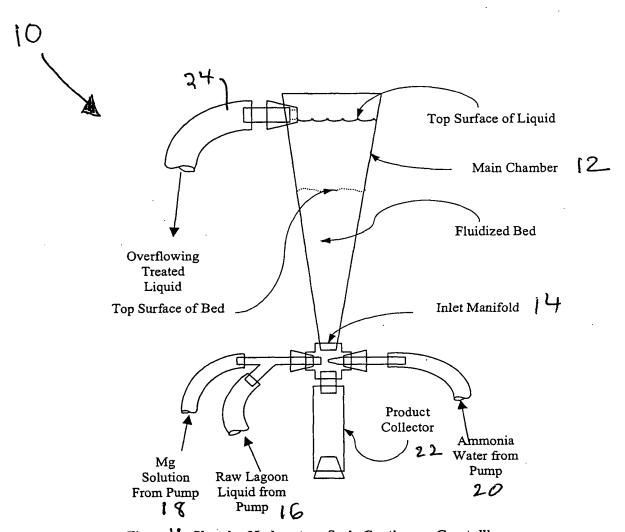
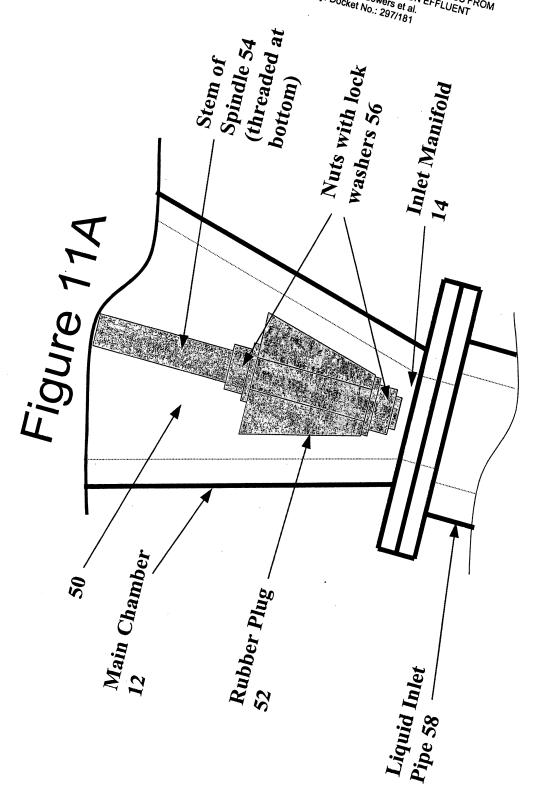


Figure | | : Sketch of Laboratory-Scale Continuous Crystallizer



Title: APPARATUS AND METHOD FOR REMOVING PHOSPHORUS FROM

WASTE LAGOON EFFLUENT

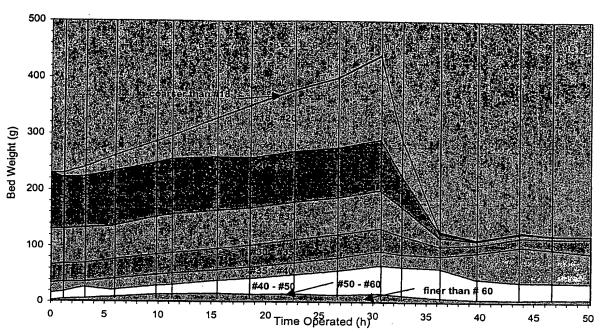


Figure 12: First Series of FCRs: Bed Weight (g), Broken Down by Particle Size (Standard Sieve), vs. Time Operated (h) (Numbered Vertical Strips Correspond with Runs)

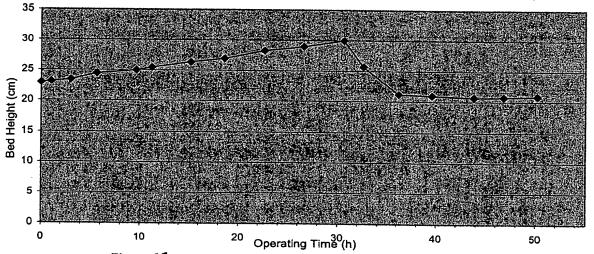


Figure [3: First Series of FCRs: Bed Height (cm) at End of Run vs. Operating Time (h)
(Run Numbers Indicated)

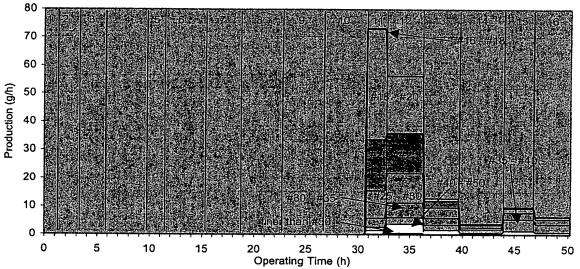


Figure 14:First Series of FCRs: Production (g/h), Averaged Over Each Run,
Broken Down by Particle Size (Standard Sieve)
(Numbered Vertical Strips Correspond with Runs)

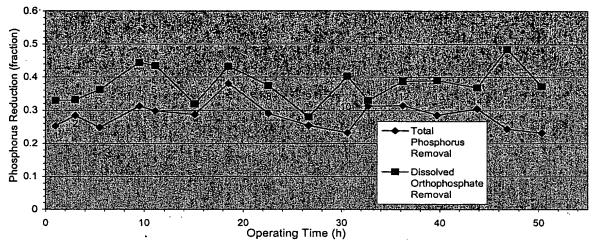


Figure 15: First Series of FCRs: Phosphorus Reduction (fraction) vs.

Operating Time (h)

(Run Numbers Indicated)

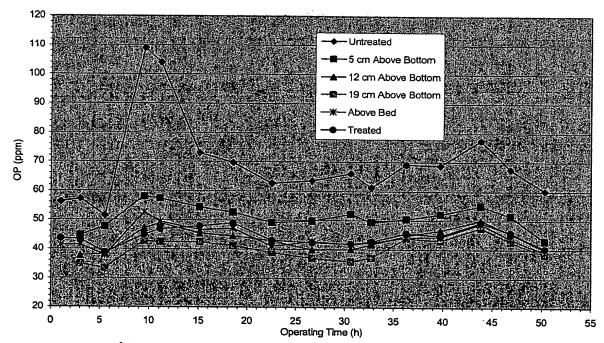


Figure (6, First Series of FCRs: OP (ppm) at Various Sampling Points vs. Operating Time (h)
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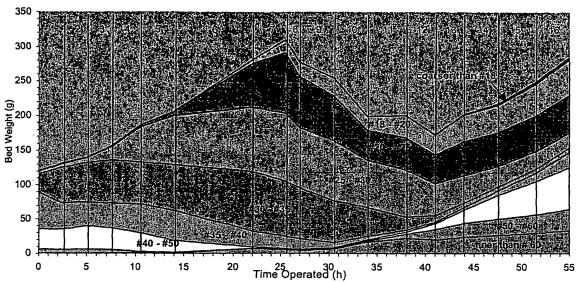


Figure 17: Second Series of FCRs: Bed Weight (g), Broken Down by Particle Size (Standard Sieve), vs. Time Operated (h) (Numbered Vertical Strips Correspond with Runs)

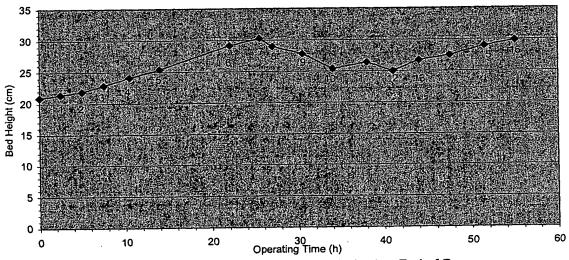


Figure |8:Second Series of FCRs: Bed Height (cm) at End of Run vs.

Operating Time (h)

(Run Numbers Indicated)

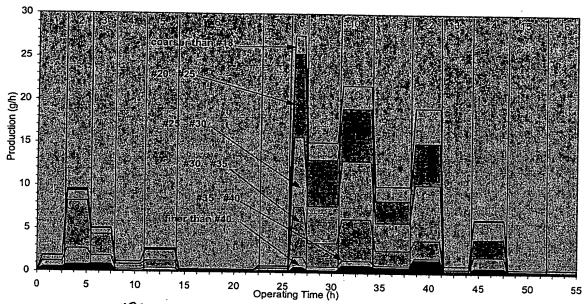


Figure 19: Second Series of FCRs: Production (g/h), Averaged Over Each Run, Broken Down by Particle Size (Standard Sieve)
(Numbered Verticle Strips Correspond with Runs)

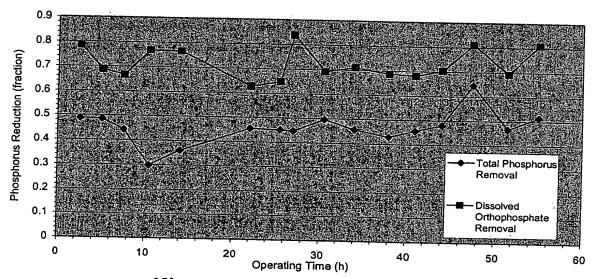


Figure 20: Second Series of FCRs: Phosphorus Reduction (fraction) vs. Operating Time (h) (Run Numbers Indicated)

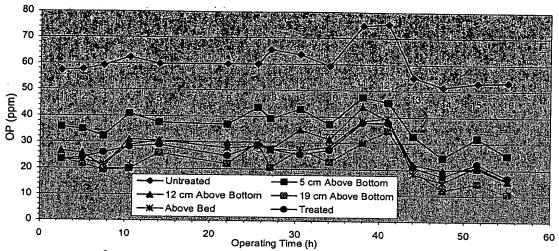


Figure 2 Second Series of FCRs: OP (ppm) at Various Sampling Points vs. Operating Time (h)
(Run Numbers Indicated)

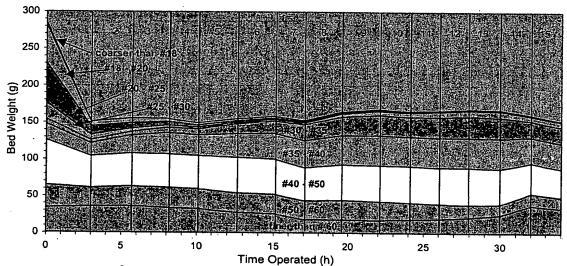


Figure 22: Third Series of FCRs: Bed Weight (g), Broken Down by Particle Size (Standard Sieve), vs. Time Operated (h) (Numbered Vertical Strips Correspond with Runs)

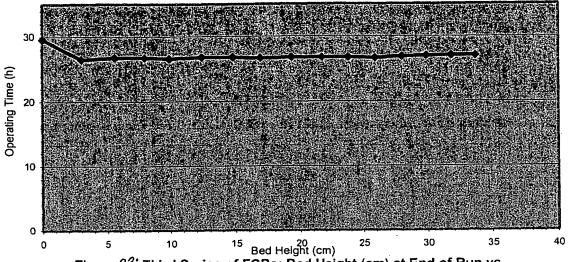


Figure 23: Third Series of FCRs: Bed Height (cm) at End of Run vs.

Operating Time (h)

(Run Numbers Indicated)

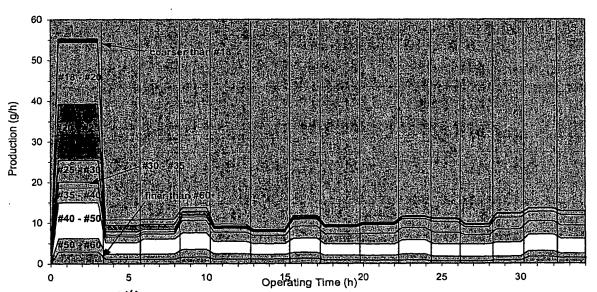


Figure 24: Third Series of FCRs: Production (g/h), Averaged Over Each Run, Broken Down by Particle Size (Standard Sieve)

(Numbered Vertical Strlp Correspond with Runs)

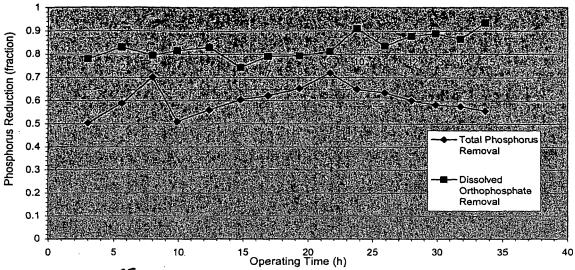


Figure 25: Third Series of FCRs: Phosphorus Reduction (fraction) vs.

Operating Time (h)

(Run Numbers Indicated)

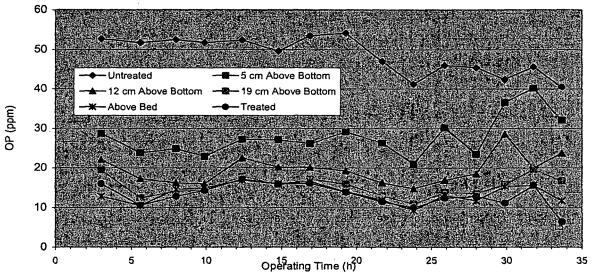


Figure 26: Third Series of FCRs: OP (ppm) at Various Sampling Points vs. Operating Time (h)
(Run Numbers Indicated)

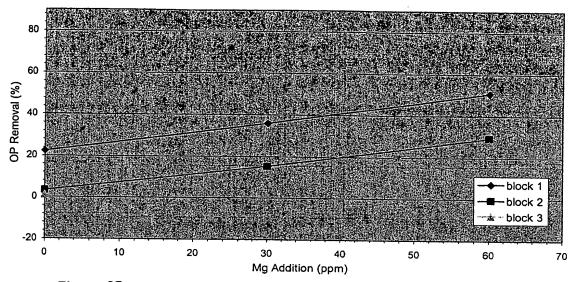


Figure 27: MVRs: OP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 41.2 L/h Flow

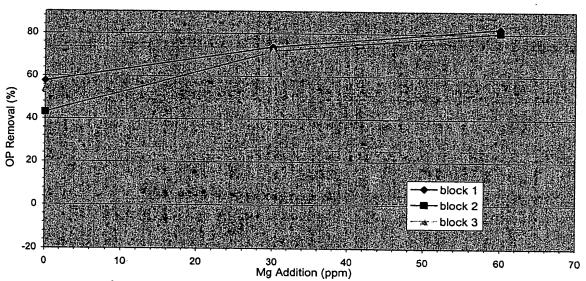


Figure 28: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as TAN) Ammonia Addition and 41.2 L/h Flow

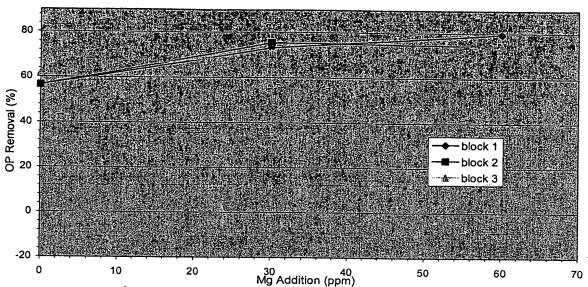


Figure 29: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 200 ppm (as TAN) Ammonia and 41.2 L/h Flow

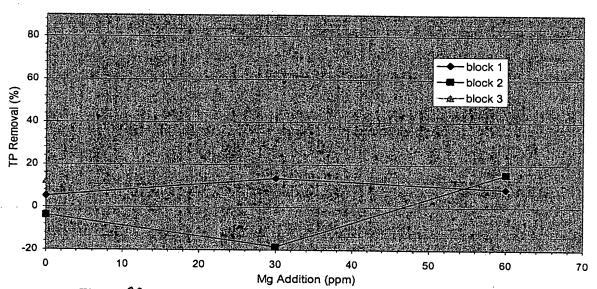


Figure 30: MVRs: TP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 41.2 L/h Flow

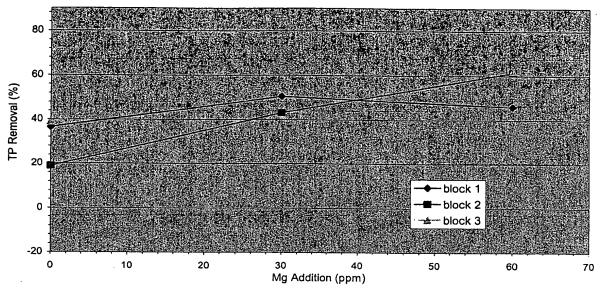
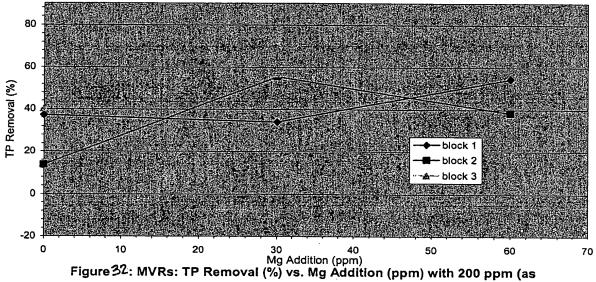
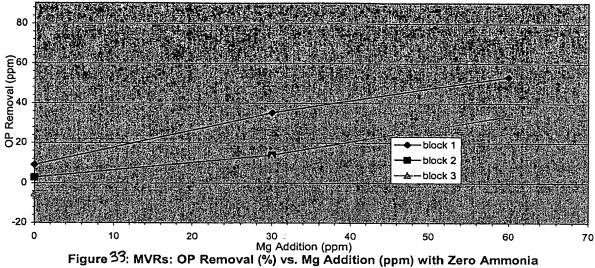


Figure 31: MVRs: TP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as TAN) Ammonia and 41.2 L/h Flow



TAN) Ammonia and 41.2 L/h Flow



and 56.8 L/h Flow

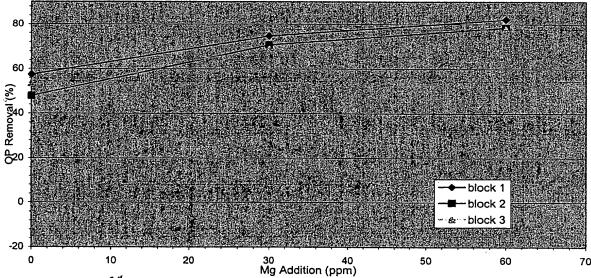
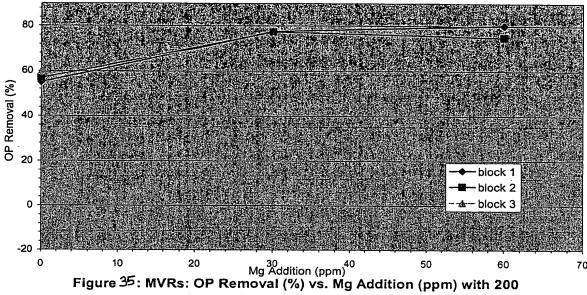


Figure 34: MVRs: OP Removal (%) vs. Mg Addition (ppm) with 100 ppm (as TAN) Ammonia and 56.8 L/h Flow



ppm (as TAN) Ammonia and 56.8 L/h Flow

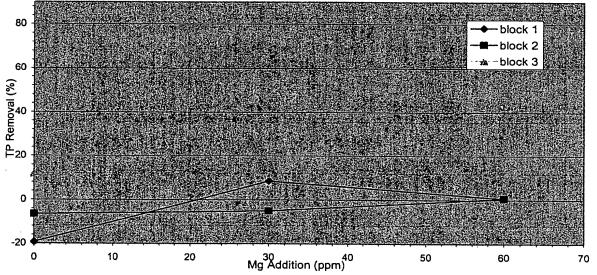
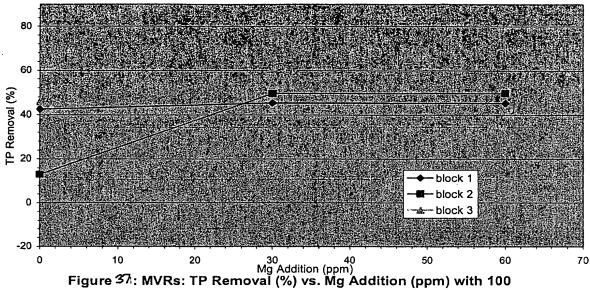
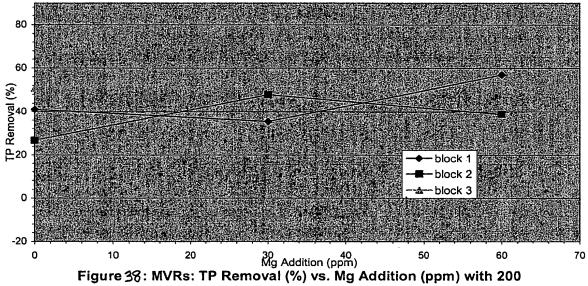


Figure 36: MVRs: TP Removal (%) vs. Mg Addition (ppm) with Zero Ammonia and 56.8 L/h Flow



ppm (as N) Ammonia and 56.8 L/h Flow



ppm (as N) Ammonia and 56.8 L/h Flow

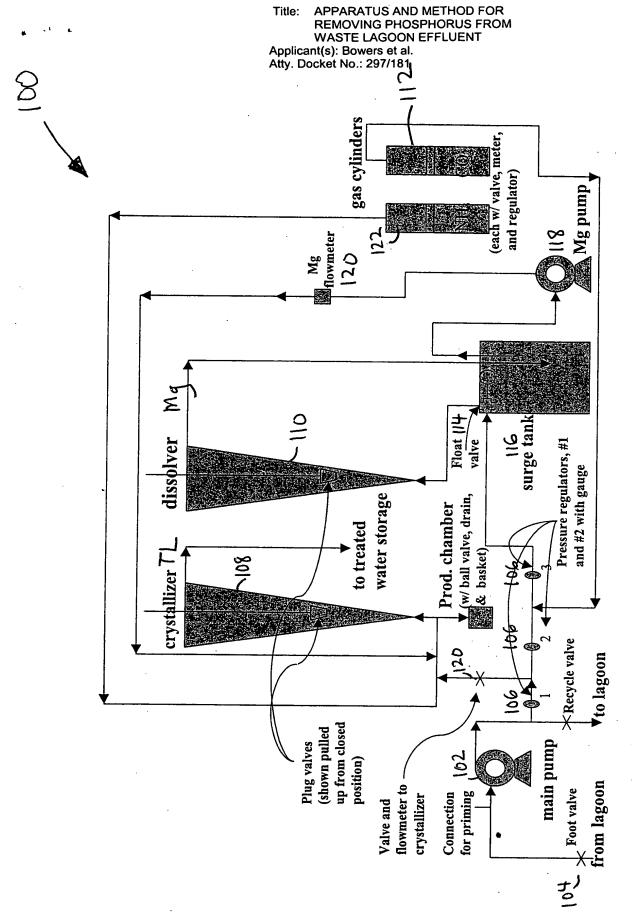


Figure 34: Schematic Representation of Field-Scale Crystallizer, Showing Principal Components

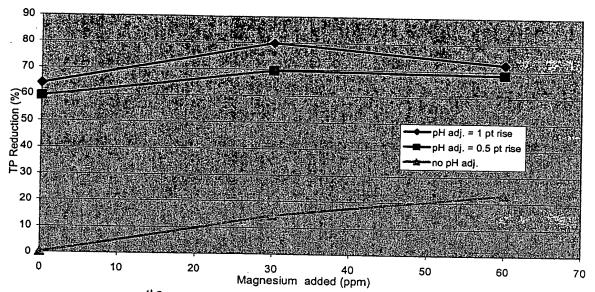


Figure 40: TP Reduction (%) vs. Magnesium added (ppm) at Lower Flow Rate (341 L/h)

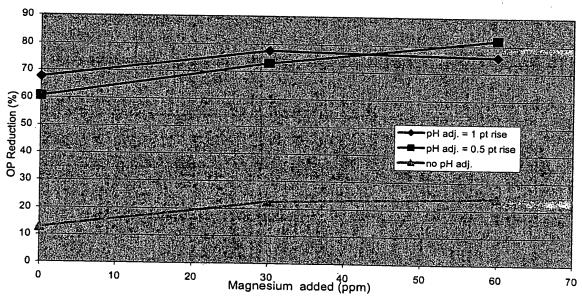


Figure 41: OP Reduction (%) vs. Magnesium added (ppm) at Lower Flow Rate (341 L/h)

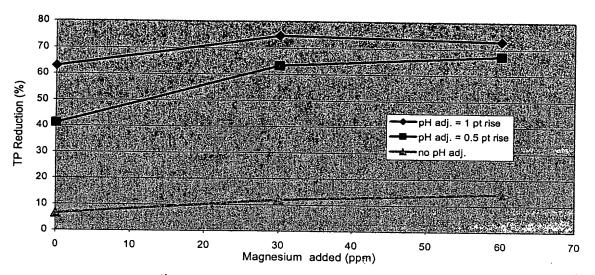


Figure 42: TP Reduction (%) vs. Magnesium added (ppm) at Higher Flow Rate (568 L/h)

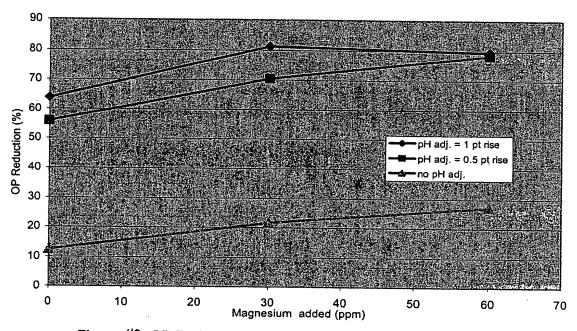


Figure 43: OP Reduction (%) vs. Magnesium added (ppm) at Higher Flow Rate (568 L/h)